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Lake Grassmere & South Island Refinery

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SAFETY DATA SHEET: BULK BRINE (Sodium Chloride)

Version: B

Revision date: 15/07/2025

Section 1: Identification of Substances and Supplier

Product name: Bulk Brine

Recommended use: Process manufacturing of Food and Pharmaceutical Salt

Manufacturer/

Supplier information: As above

Telephone number: +64 7 575 6193 Head Office Mount Maunganui (New Zealand)

Emergency telephone Outside New Zealand: +64 3 35 30199

Number: Within New Zealand: 0800 CHEMCALL ® (0800 243 62255)

Section 2: Hazards identification

HSNO Approval Code: Does not have an individual approval but may be used under an appropriate

group standard

HSNO Classification: H319: causes serious eye irritation, Classification eye Irrit.2

GHS Classification: Not classified as hazardous under GHS criteria

GHS Label Elements: No hazard pictogram, no signal word, no hazards statement

Precautionary statements: No precautionary statements required

Hazards Statements: Non-hazardous material under normal handling

and storage conditions.

Section 3: Classification of Substance

Chemical Name: Naturally produced aqueous solution of substantially Sodium Chlorides

but containing other inorganic salt found in sea water

CAS Number: 7647-14-5 **EC Number:** 231-598-3

Section 4: First Aid Measures

Inhalation: Not normally a risk, but some discomfort may follow where working

with dusty product. If inhaled, remove person to fresh air. If symptoms

persist, seek medical attention. Allow to drink plenty of water.

Ingestion: Rinse mouth. If large amounts are ingested, obtain immediate

medical attention (especially if vomiting has not occurred).

Eye Contact: Rinse immediately with plenty of water, also under the eyelids, for at

least 20 minutes. If irritation occurs, seek medical attention.

Skin Contact: Wash with plenty of water. If irritation persist, seek medical attention.

Workplace facilities: Emergency showers and eye wash recommended.



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NOTES FOR MEDICAL PERSONNEL

Swallowed: Give water to drink. No need to induce vomiting. **Eye:** Irrigate with copious quantities of slow flowing

water for minimum of 20 minutes. Eyelids to be

held open.

Skin: Brush off clothing and wash skin thoroughly with plenty of water.

Inhaled: Not normally a risk but some may experience some discomfort if

working with dusty product. If exposure has occurred allow the victim to

drink water

Section 5: Firefighting measures

Extinguishing media: Use extinguishing measures that are suitable for the surrounding fire and

the environment

No limitations of extinguishing are given for this substance

Sodium chloride is not flammable and does not support combustion.

Special hazards arising

from the substance or mixture: Hydrogen chloride gas, Sodium oxides, ambient fire may liberate

hazardous vapours

Advise for fire fighters: Wear self-contained breathing apparatus and protective clothing if

necessary

Section 6: Accidental release measures

Personal Precautions: Ensure adequate ventilation. Avoid contact with eyes and skin.

Environmental Precautions: Avoid uncontrolled releases in the environment.

Methods for clean up: Cover drains. Flush area to dissolve in sufficient amounts of water to

meet existing water quality standards.

Section 7: Handling and Storage

No particular requirements.

Section 8: Exposure Controls/ Personal Protection

Subsection 1: Workplace Exposure Guidelines

Occupational Exposure: As total dust 10mg/m3 (8hr TWA)

Limits: As respirable dust 4mg/m3 (8hr TWA)

Dangerous Exposure: Non specified.

Engineering Controls: Static electricity can be generated by pneumatic conveying, therefore

pipes should be bonded and earthed, especially in environments where a spark could prove hazardous. Structural integrity of various metals used in equipment and structures should be regularly checked as it may accelerate the corrosion of common metals (especially

aluminium).



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Subsection 2: Engineering Controls

Respiratory Protection: If the process is such that salt dust is generated, an approved

respirator should be worn.

Hand Protection: Gloves to be worn if prolonged contact is anticipated.

Dry salt and concentrated solutions can cause withdrawal of fluid

from the skin.

Eye Protection: Wear chemical safety goggles in situations where contact with the

eyes may occur.

Skin Protection: Skin should be washed to remove salt.

Dry salt and concentrated solutions can cause withdrawal of fluid

from the skin.

Other Protective Measures: An eyewash and hand washing facilities should be readily available.

Section 9: Physical and Chemical Properties

Appearance: Liquid

Colour:

Boiling Point:

Flammability:

Flash Point:

Explosive Properties:

Oxidising Properties:

Vapour Pressure:

Clear to hazy white

>100°C (>212 °F)

Not available

Not available

Not available

Not available

Water solubility: Slightly soluble in water

Viscosity & Vapour Density: Not applicable

Section 10: Stability and Reactivity

Density:

Chemical Stability: Stable

Conditions to avoid: Reacts with strong sulphuric acid or nitric acid to give

Not available

hydrogen chloride gas

Material to avoid: Under wet conditions can corrode many common metals, particularly

iron, aluminum and zinc. Stainless steel and monel resist attack.

Hazard Decomposition: Trace amounts of hydrogen chloride gas may be evolved at

in excess of 800°C. Contains no water of crystallization. Does not

react with alkalis at ordinary temperatures.



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Section 11: Toxicological Information

Eyes: Dust may be irritating

Skin: Irritation after prolonged contact

Ingestion: Salt is an essential constituent of the diet. It provides important body

electrolytes and is the source of hydrochloric acid present in the gastric juices. The blood stream contains nearly 1% sodium chloride.

In normal industrial use salt is non-hazardous.

LD50 3000mg/kg oral, rat.

Inhalation: Dusts may be irritating

Carcinogenity: Not considered to be a carcinogen

Mutagenity: Not considered to be a mutagen

Reproductive Effects: Not identified

Section 12: Ecological Information

Ecotoxicity: Do not empty into drains. A maximum value of 412 mg/l ensures the protection of all aquatic life. Source: Water Research Centre – September 1990.

96 hour LC 50 (Fish) 6750 mg/l 48 hour EC 50 (Daphnia) 2024 mg/l

72 hour IC 50 (Algae) 3014 mg/l Daphnia Sub acute 1062 mg/l Fish Subacute 433

mg/l

BOD 5 day 0 mg/l COD 0 mg/l

Earthworm Toxicity 1000 hg/cm2

Persistence/Degradability/Bioaccumulation: No information available

Section 13: Disposal Considerations

Spills: Collect solid salt in a conventional manner, wash the spill area down with water

if necessary.

Disposal methods: Refer to the Local council bylaws and Land Waste Management Authority.

Dissolved material in excess water is normally suitable for disposal in storm water

system.

Section 14: Transport Information

Material is not included in the requirements for "Transport of Dangerous Goods on Land"

UN Number: Not regulated

Transport Hazards Class (EEC Classification): Not classified as hazardous for transport



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Section 15: Regulatory Information

HSNO Classification: Not classified as hazardous

HSNO Approval Code: Exempt from HSNO approval

GHS Label Elements: Not applicable. Non-hazardous substance

Section 16: Other Information

Disclaimer:

The information provided in this SDS is accurate to the best of our current knowledge of the chemical substance at the date of issue. It does not represent any guarantee of the properties of the product. Dominion Salt shall not be held liable for any damage resulting from handling or from contact with above product and the user assumes all responsibility for handling and use according to laws and regulations.

This document has been prepared in compliance with New Zealand's HSNO regulations and the Globally Harmonized System (GHS) of Classification and Labelling of Chemicals.